



Video Solution on Website:-

<https://physicsaholics.com/home/courseDetails/88>

Video Solution on YouTube:-

<https://youtu.be/iP4yeRrnKgl>

Written Solution on Website:-

<https://physicsaholics.com/note/notesDetails/28>

- Q 1. A material whose K absorption edge is 0.15 \AA is irradiated with 0.1 \AA X-rays. The maximum kinetic energy of photoelectrons that are emitted from K-shell is –
(a) 41 KeV (b) 51 KeV (c) 61 KeV (d) 71 KeV
- Q 2. For a given material, the energy and wavelength of characteristic x-ray satisfy–
(a) $E(K\alpha) > E(K\beta) > E(K\gamma)$
(b) $E(M\alpha) > E(L\alpha) > E(K\alpha)$
(c) $\lambda(K\alpha) > \lambda(K\beta) > \lambda(K\gamma)$
(d) $\lambda(M\alpha) > \lambda(L\alpha) > \lambda(K\alpha)$

COMPREHENSION (Q3 to Q5)

A tungsten target ($z = 74$) is bombarded by electrons in an X-ray tube. The K, L and M atomic X-rays energy levels for tungsten are 69.5, 11.3 and 2.3 keV

- Q 3. The minimum value of the accelerating potential that will permit the production of the characteristic $k\beta$ and $k\alpha$ lines of tungsten–
(a) 69.5 kV (b) 58.2 kV (c) 67.2 kV (d) 11.3 Kv
- Q 4. For the same accelerating potential, what is λ_{min} ?
(a) 16.9 pm (b) 17.9 pm (c) 18.9 pm (d) 19.9 pm
- Q 5. What is $k\beta$ wavelength ?
(a) 16.5 pm (b) 17.5 pm (c) 18.5 pm (d) 21.5 pm
- Q 6. $K\alpha$ wavelength emitted by an atom of atomic number $Z = 11$ is λ . Find the atomic number for an atom that emits $K\alpha$ radiation with wavelength 4λ :
(a) $Z = 6$ (b) $Z = 4$ (c) $Z = 11$ (d) $Z = 44$
- Q 7. X-rays will not show the phenomenon of :
(a) diffraction (b) polarisation
(c) deflection by electric field (d) interference
- Q 8. The wavelength of $K\alpha$ X-rays produced by an X-ray tube is 0.76 \AA . The atomic number of the anticathode material is
(a) 82 (b) 41 (c) 20 (d) 10



- Q 9. The minimum wavelength of X-ray that can be produced in a Coolidge tube depends on
- the metal used as the target
 - the intensity of the electron beam striking the target
 - the current flowing through the filament
 - the potential difference between the cathode and the anode
- Q 10. If the potential difference applied across a Coolidge tube is increased,
- the wavelength of the K_{α} line will increase
 - the wavelength of the K_{β} line will decrease
 - the difference in wavelength between the K_{α} and K_{β} lines will decrease
 - none of the above
- Q 11. If the voltage applied to a X-ray tube is being increased $\eta = 1.5$ times, the short wave limit of the X-ray continuous spectrum shifts by $\Delta\lambda = 26$ pm. Find the initial voltage applied to the tube.
- 10 KV
 - 12 KV
 - 14 KV
 - 16 KV
- Q 12. The potential difference applied to an X-ray tube is 5 kV and the current through it is 3.2 mA. Then the number of electrons striking the target per second is
- 2×10^{16}
 - 5×10^6
 - 1×10^{17}
 - 4×10^{15}
- Q 13. When the voltage applied to an X-ray tube is increased from $V_1 = 10$ kV to $V_2 = 20$ kV, the wavelength interval between the K_{α} line and the short-wave cut off of the continuous X-ray spectrum increases by a factor $\eta = 3.0$. Find the atomic number of the element of which the tube's anticathode is made.
- 11
 - 20
 - 29
 - 38
- Q 14. If a potential difference of 20,000 volts is applied across an X-ray tube, the cut-off wavelength will be
- 6.21×10^{-10} m
 - 6.21×10^{-11} m
 - 6.21×10^{-12} m
 - 3.1×10^{-11} m
- Q 15. Which of the following pairs constitute very similar radiations?
- Hard ultraviolet rays and soft X-rays
 - Soft ultraviolet rays and hard X-rays
 - Very hard X-rays and low-frequency γ -rays
 - Soft X-rays and γ -rays
- Q 16. When an electron moving at a high speed strikes a metal surface, which of the following are possible?
- The entire energy of the electron may be converted into an X-ray photon.



- (b) Any fraction of the energy of the electron may be converted into an X-ray photon.
- (c) The entire energy of the electron may get converted to heat.
- (d) The electron may undergo elastic collision with the metal surface

- Q 17. The wavelength of K_{α} X-rays for lead isotopes Pb^{208} , Pb^{206} , Pb^{204} are λ_1 , λ_2 and λ_3 respectively. Then :
- (a) $\lambda_1 = \lambda_2 = \lambda_3$
 - (b) $\lambda_1 > \lambda_2 > \lambda_3$
 - (c) $\lambda_1 < \lambda_2 < \lambda_3$
 - (d) $\lambda_2 = \sqrt{\lambda_1 \lambda_3}$

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Answer Key

Q.1 a	Q.2 c,d	Q.3 a	Q.4 b	Q.5 c
Q.6 a	Q.7 c	Q.8 b	Q.9 d	Q.10 d
Q.11 d	Q.12 a	Q.13 c	Q.14 b	Q.15 a,c
Q.16 a,b,c	Q.17 a,d			